

Production of Xe standards for the calibration of noble gas sampler stations and laboratory equipment

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Outline

- Motivation
- Upgraded Xenon production facility at Jyväskylä University, Finland
- Foil-to-gas bottle conversion at STUK
- Expected performance of the technique
- Conclusions

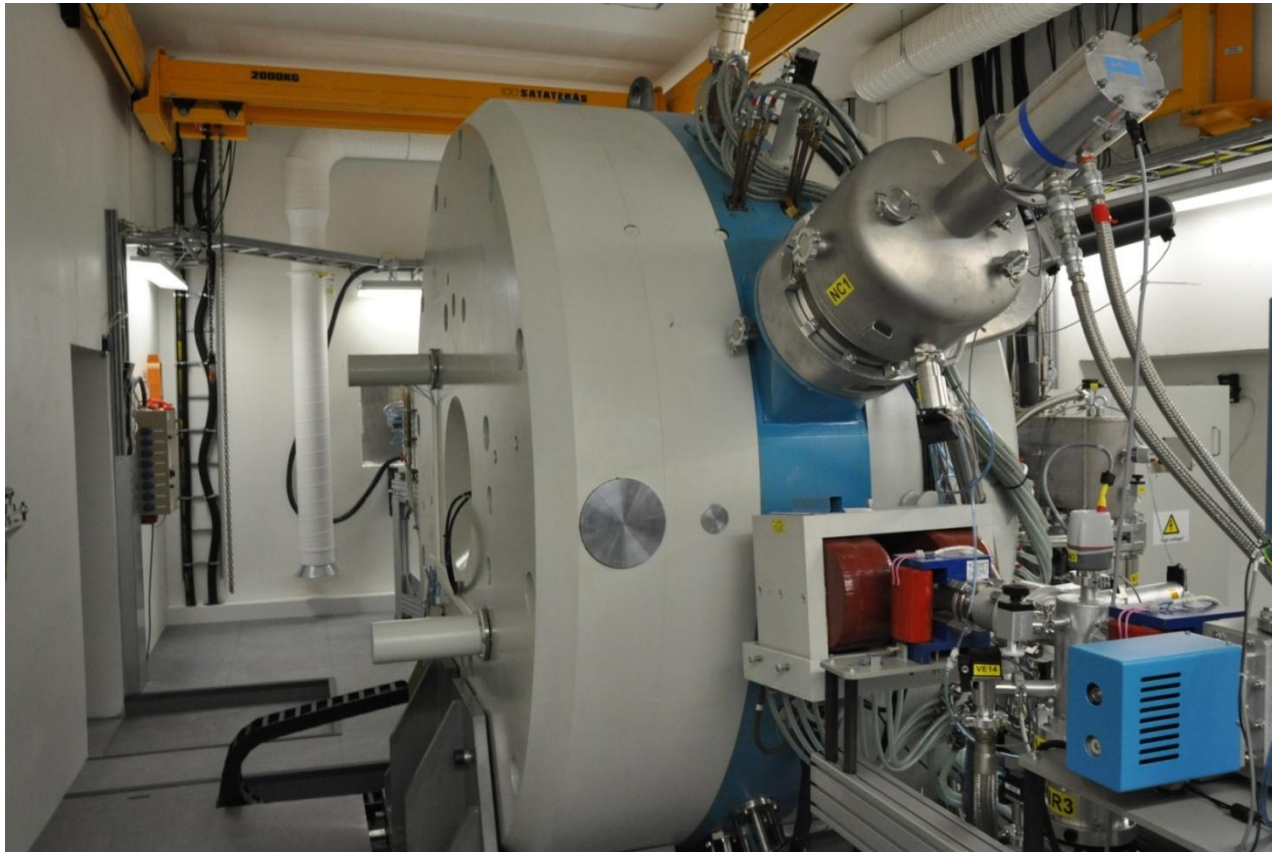
Motivation

- Pure Xe samples are required for equipment calibration and software performance evaluation.
- $^{133\text{m}}\text{Xe}/^{133}\text{Xe}$ ratio is a key signature related to discrimination between nuclear test explosion and reactor releases.

$$\text{Mass}(^{133\text{m}}\text{Xe}) = \text{Mass}(^{133}\text{Xe}) + \underbrace{233 \text{ keV}}_{\sim 1/2 \text{ electron mass}}$$

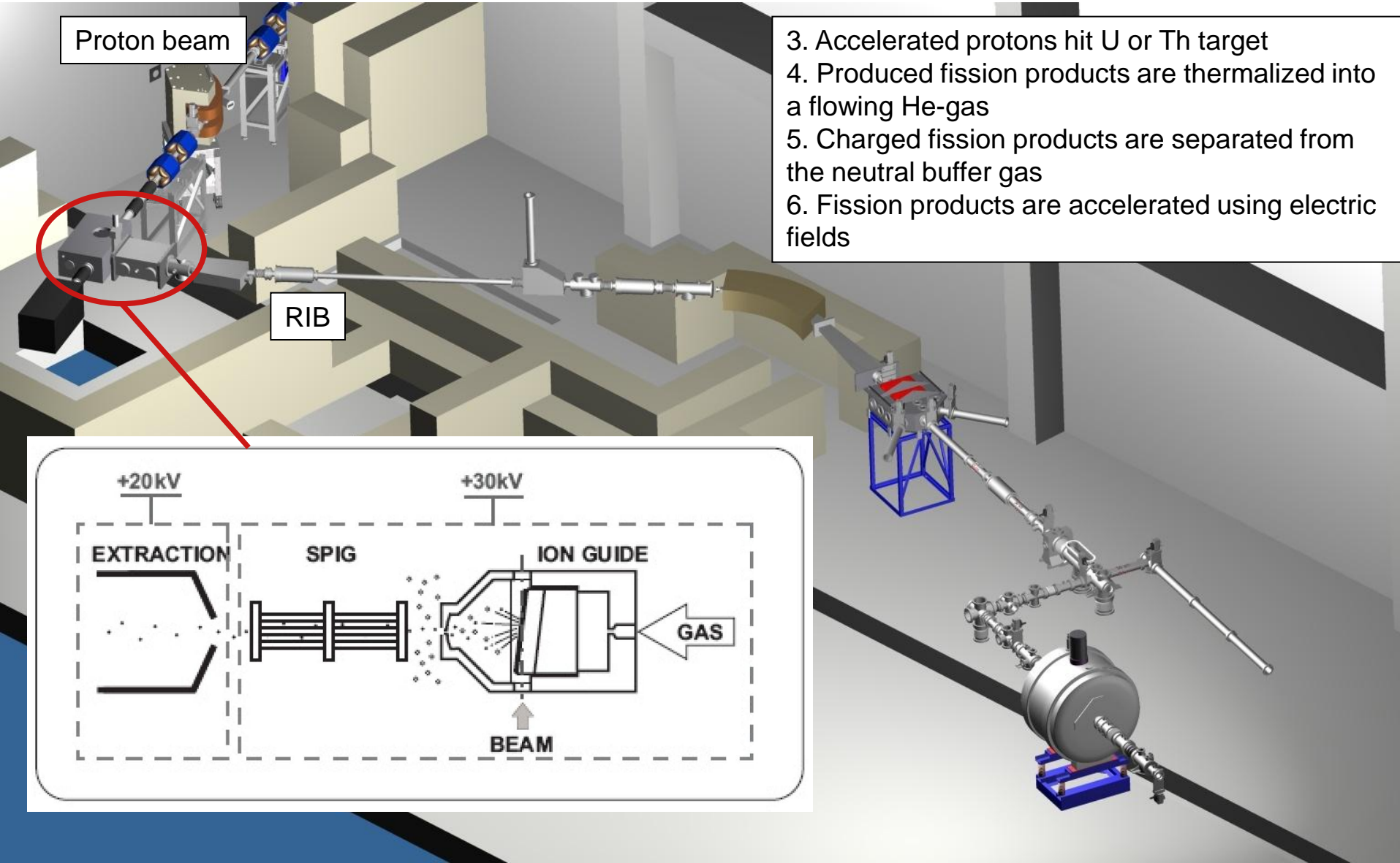
New cyclotron in Jyväskylä

1. Proton beam is produced with the plasma ion source
2. Beam is accelerated using the cyclotron that will primarily serve the upgraded IGISOL/JYFLTRAP facility



Production of Radioactive Ion Beam (RIB)

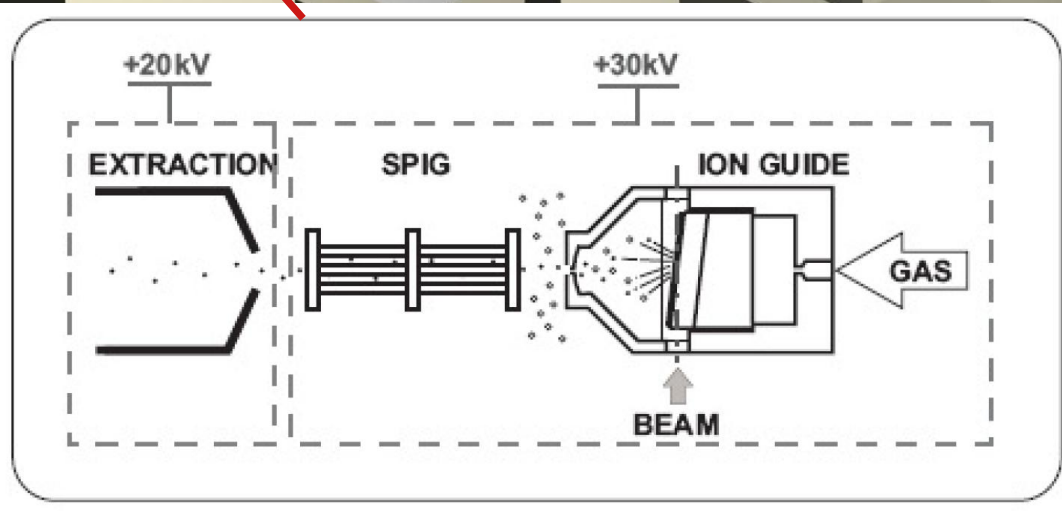
Upgraded IGISOL/JYFLTRAP facility



Proton beam

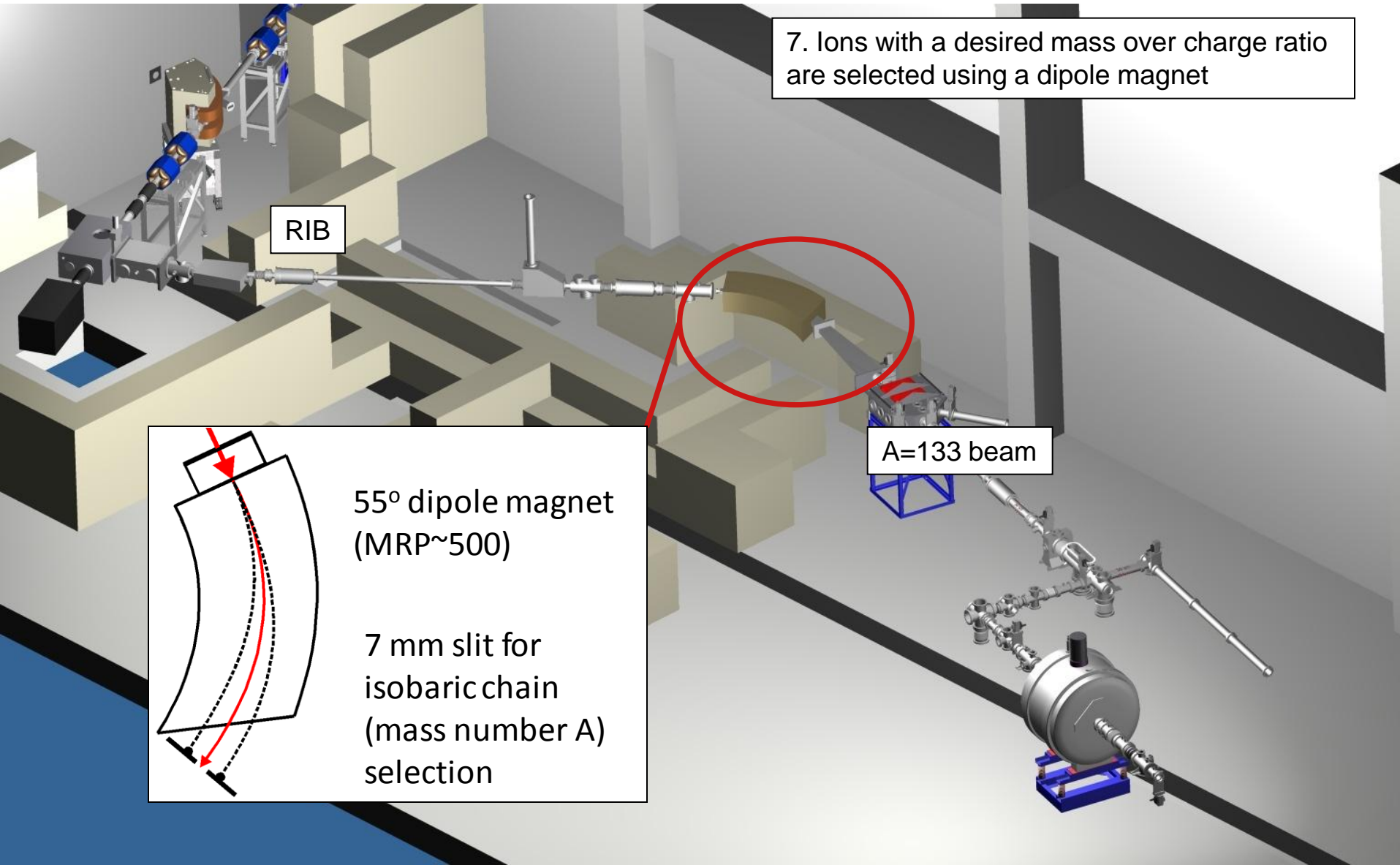
RIB

3. Accelerated protons hit U or Th target
4. Produced fission products are thermalized into a flowing He-gas
5. Charged fission products are separated from the neutral buffer gas
6. Fission products are accelerated using electric fields



Rough mass separation

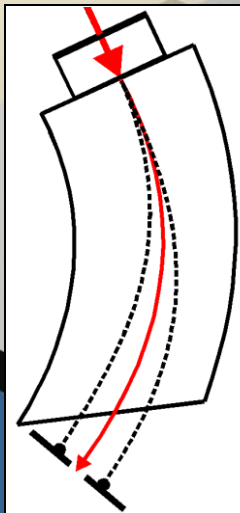
Upgraded IGISOL/JYFLTRAP facility



7. Ions with a desired mass over charge ratio are selected using a dipole magnet

RIB

A=133 beam

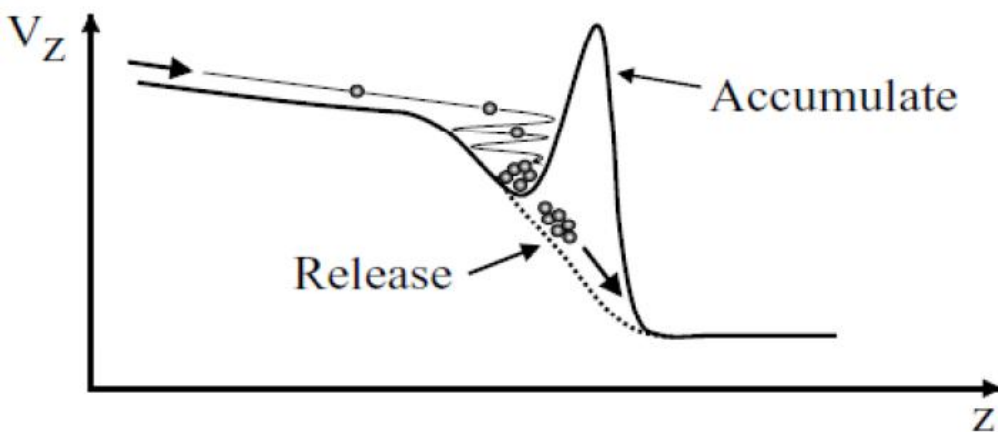
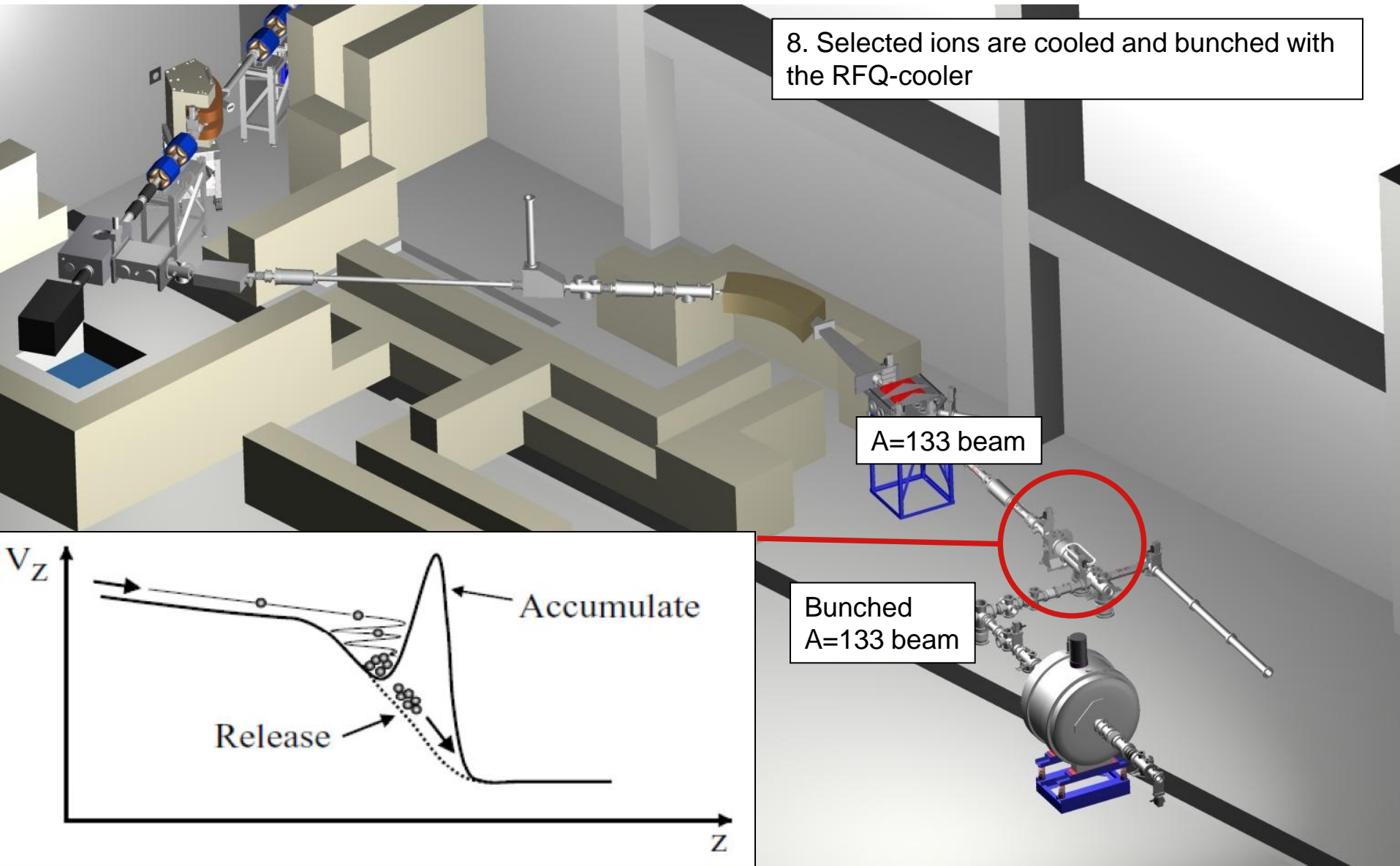


55° dipole magnet
(MRP~500)

7 mm slit for
isobaric chain
(mass number A)
selection

Cooling and bunching of the beam

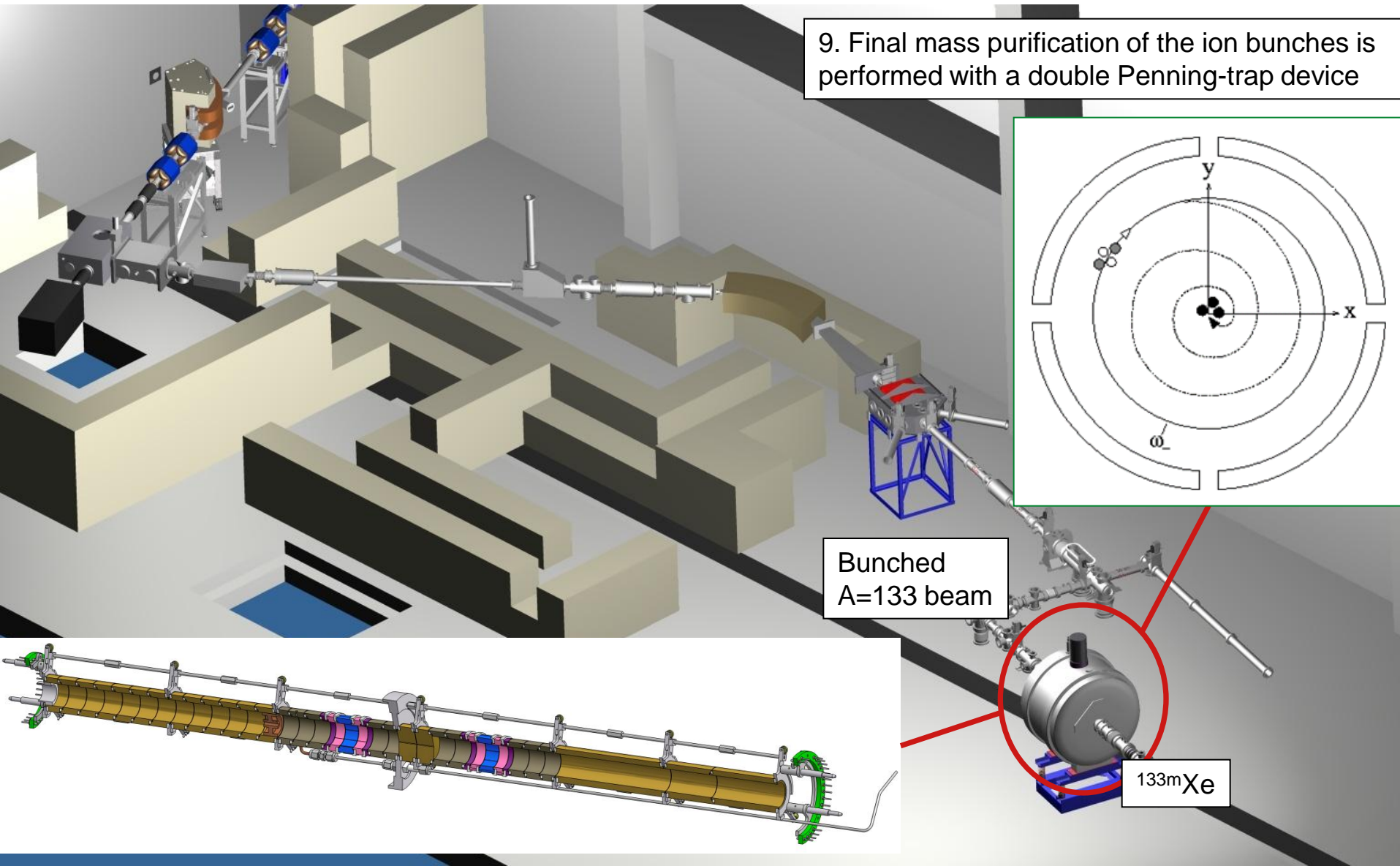
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Final mass purification

Upgraded IGISOL/JYFLTRAP facility

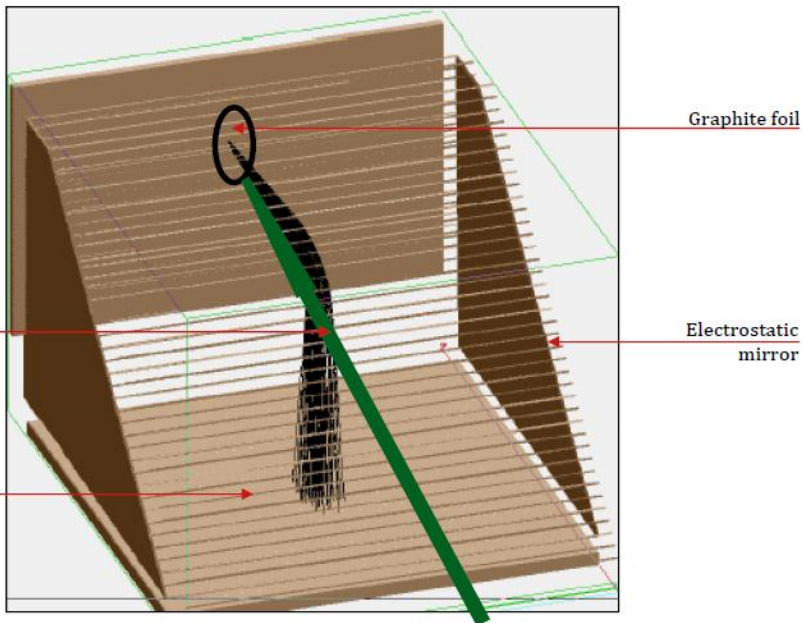
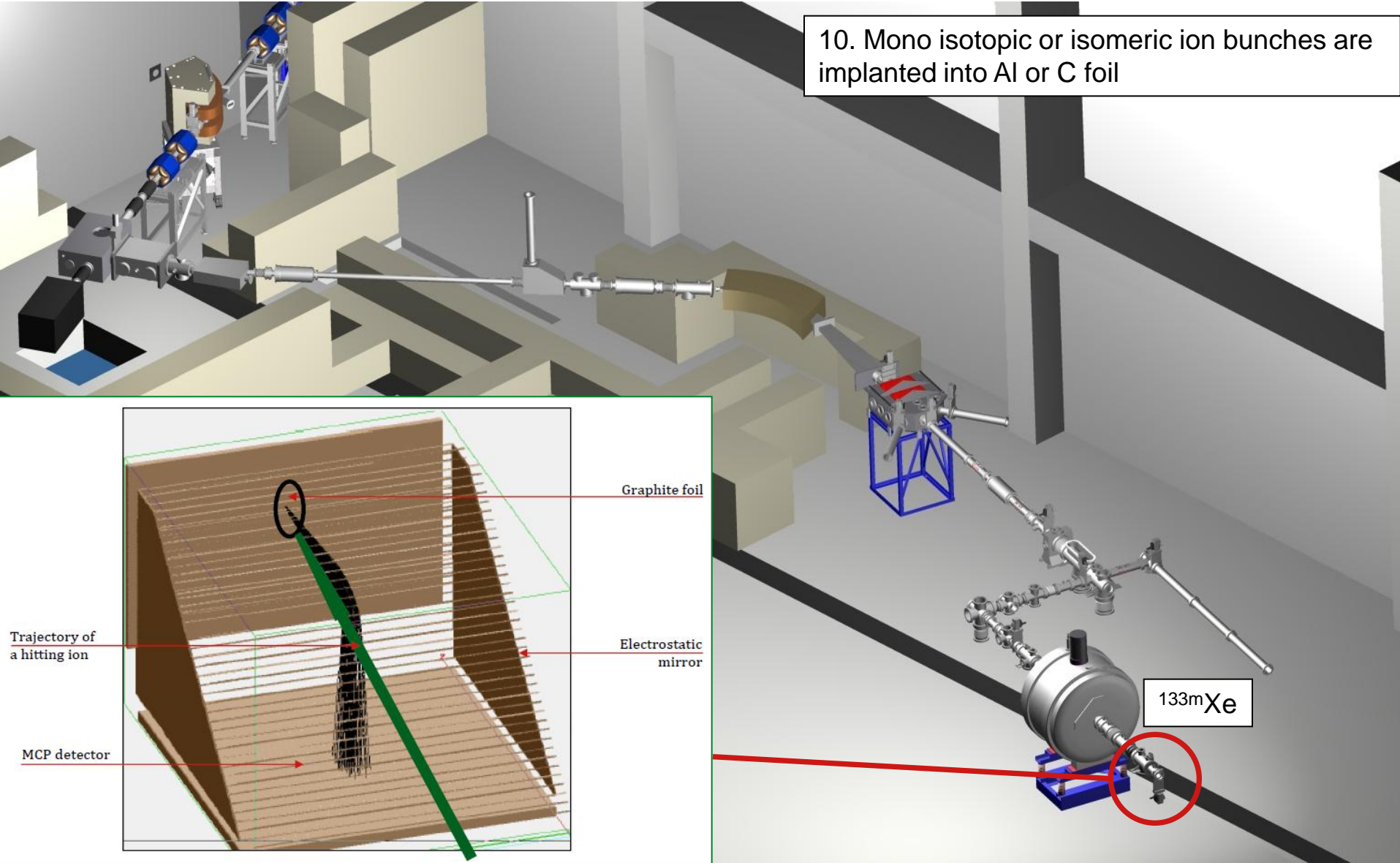
9. Final mass purification of the ion bunches is performed with a double Penning-trap device



Implantation to solid catcher

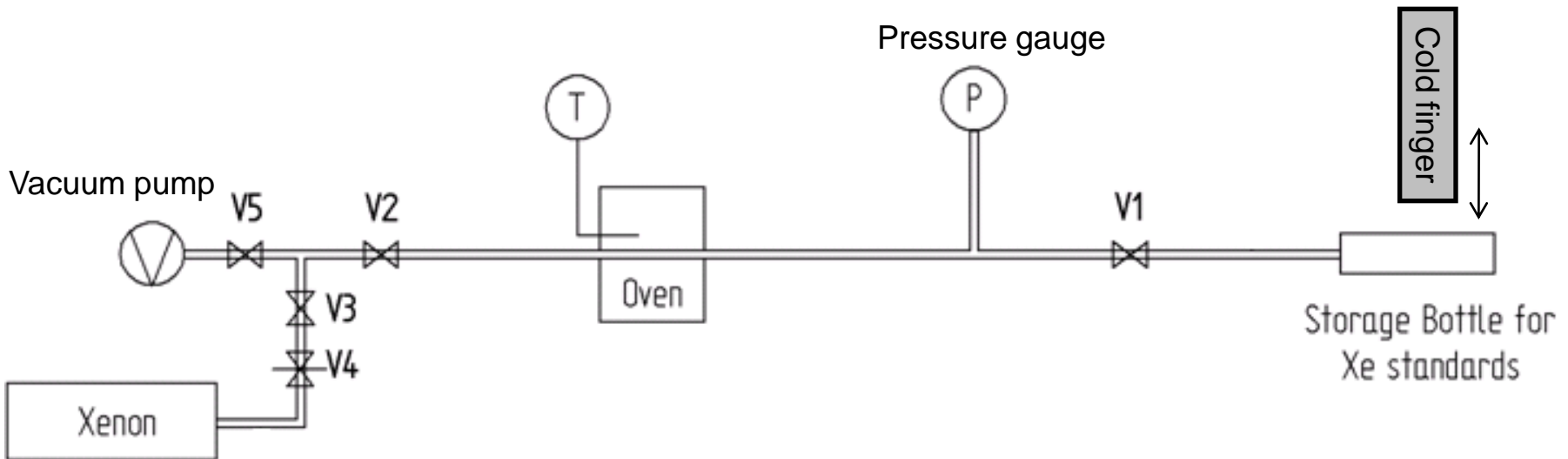
Upgraded IGISOL/JYFLTRAP facility

10. Mono isotopic or isomeric ion bunches are implanted into Al or C foil



Foil-to-gas bottle conversion at STUK

11. Gamma spectrometry for actual and reference samples
12. Implanted species are transferred from the foil to a gas bottle.



Graphite foil thickness used is $215 \mu\text{g}/\text{cm}^2$.

2 h at $400 \text{ }^\circ\text{C}$ is enough to release the implanted Xe completely.

Melting point of Xe is $-111.9 \text{ }^\circ\text{C}$ while liquid nitrogen temperature is $-195.8 \text{ }^\circ\text{C}$.

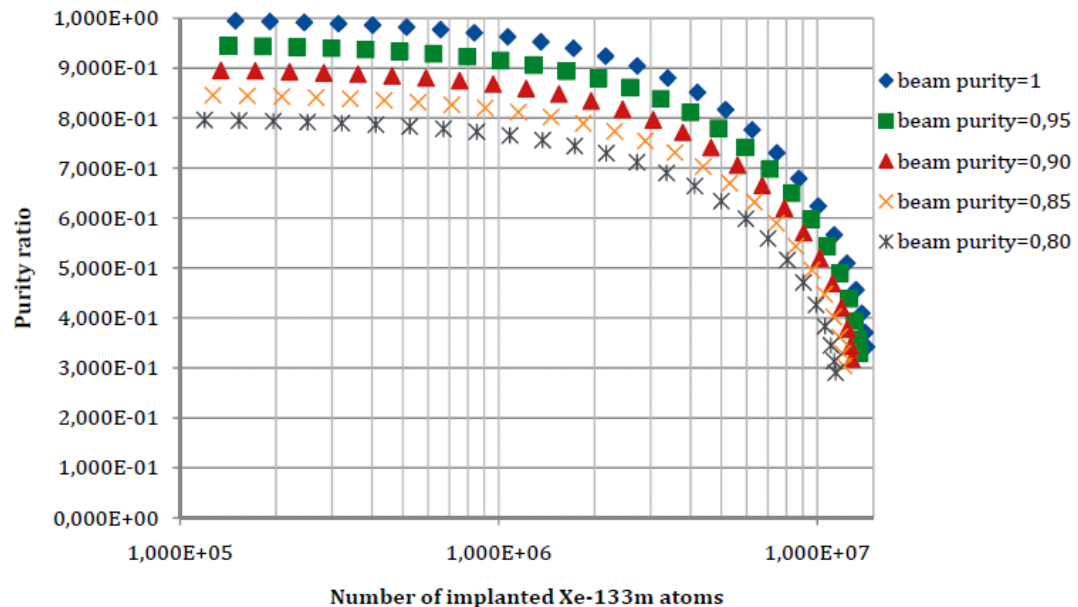
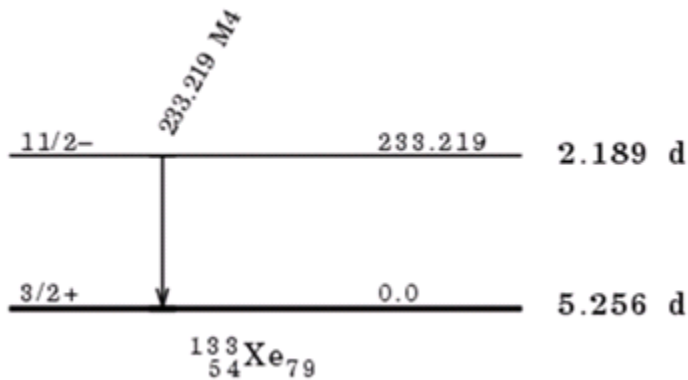
Expected performance of the technique

Using old IGISOL/JYFLTRAP facility it has been shown that pure samples of $^{131\text{m}}\text{Xe}$, $^{133\text{m}}\text{Xe}$, ^{133}Xe and ^{135}Xe can be prepared with the rate of some tens of ions per second.

Example, $^{133\text{m}}\text{Xe}$ sample purity at the end of implantation

Assume a beam purity, $^{133\text{m}}\text{Xe}/(^{133\text{m}}\text{Xe}+^{133}\text{Xe})$, of 1 and a beam intensity of 50 ions/s.

These figures lead to 0.965 sample purity at the end of implantation of 10^6 $^{133\text{m}}\text{Xe}$ ions. The corresponding implantation time and $^{133\text{m}}\text{Xe}$ activity are ~ 5.5 h and ~ 3.5 Bq.



Conclusions

- Mass separator-based production of pure $^{131\text{m}}$, $^{133\text{m}}$, 133 , ^{135}Xe samples is possible
- Graphite is a good implantation foil choice for Xe
- New cyclotron primarily serving the upgraded IGISOL/JYFLTRAP facility offers much more flexibility to scheduling of experiments
- Production of Xe samples should be possible in late 2011 or early 2012

Collaboration

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